

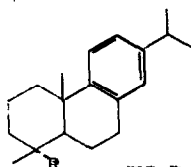
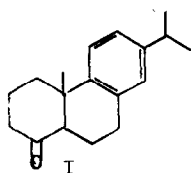
DEAMINATION OF 19-NORABIETA-8,11,13-TRIEIN-4-AMINE

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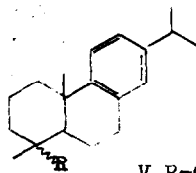
18,19-Dinorabieta-8,11,13-trien-4-one (I), a valuable intermediate in the conversion of the resin acid abieta-8,11,13-trien-19-oic acid (II) into compounds suitable for the synthesis of steroids, has been prepared in varying success.³⁻⁸ The exocyclic alkene, 19-norabieta-4(18),8,11,13-tetraene (VII), is an important precursor to the formation of this ketone (I).



II R=COOH

III R=NCO

IV R=NH₂



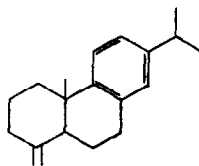
V R=OH

VI R=OAc

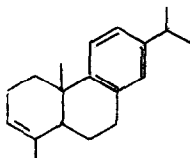
We wish to report an alternative high-yield route to I via the nitrous acid deamination of 19-norabieta-8,11,13-trien-4-amine (IV). Zeiss and Martin⁵ have reported that treatment of the C-4 isocyanate (III) with mineral acid did not give satisfactory yields of the amine (IV). In this present work however, we have found that hydrolysis of a cold (0°) ethereal solution of III with 85% (w/w) sulphuric acid resulted in the formation of the tertiary amine (IV, 73%), picrate m.p. 218-220° (lit.,⁷ 222-223°). $\nu_{\text{max}}^{\text{CCL}_4}$ 3300, 1620 cm.⁻¹ (N-H). δ (CDCl₃) 1.12 (C₁₈-Me), 1.17 (C₂₀-Me), 1.23 (doublet, J=7 c/s, (Me)₂CH-), 1.75 (NH₂), 6.94-7.31 (aromatics). Reaction of IV with saturated sodium nitrite in 5% (v/v) acetic acid at 4° for 24 hr gave: 19-norabieta-8,11,13-trien-4-ol (V, 10%), m.p. 91.5-92.5° (from light petroleum), $\nu_{\text{max}}^{\text{KBr}}$ 3425, 1136 cm.⁻¹ (C-H), δ (CCl₄) 1.07 (C-H)(D₂O), 1.12 (C₁₈-Me), 1.17 (C₂₀-Me), 1.21 (doublet, J= 7c/s, (Me)₂CH-), 6.73-7.23 (aromatics); 19-norabieta-8,11,13-trien-4-ol acetate (VI, 6%), $\nu_{\text{max}}^{\text{CS}_2}$ 1745, 1250 cm.⁻¹ (acetate); and an alkene mixture (45%).

NMR analysis of the mixture of alkenes indicated the presence of: 65% of the $\Delta^{4(18)}$ isomer (VII), $\nu_{\text{max}}^{\text{CS}_2}$ 3100, 890 cm.⁻¹ (=CH₂), δ (CCl₄) 0.99 (C₂₀

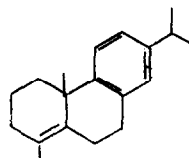
Me), 1.22 (doublet, $J=7$ c/s, $(\text{Me})_2\text{CH}-$), 4.59, 4.84 ($=\text{CH}_2$), 6.80-7.30 (aromatics); 26% of the Δ^3 alkene (VIII), δ (CCl_4) 1.03 ($\text{C}_{20}-\text{Me}$), 1.22 (doublet, $J=$



VII



VIII



IX

7 c/s, $(\text{Me})_2\text{CH}-$), 1.34 ($\text{C}_{18}-\text{Me}$), 5.45 (C_3-H); and 9% of the Δ^4 isomer (IX).

Oxidation of the hydrocarbon mixture with OsO_4 gave I (75%). $\nu_{\text{max}}^{\text{CCl}_4}$ 1718 cm^{-1} ($\text{C}=\text{O}$). δ (CCl_4) 1.02 ($\text{C}_{20}-\text{Me}$), 1.21 (doublet, $J=7$ c/s, $(\text{Me})_2\text{CH}-$) 6.76-7.35 (aromatics).

The deamination products obtained from the equatorial amine (IV) show little resemblance to those resulting from the action of nitrous acid on the aminosteroids^{9,10} aminodecalins¹¹ and the aminohydrindanes.^{12,13}

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